

The Mo-Ni-Si (Molybdenum-Nickel-Silicon) System

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Introduction

The Mo-Ni-Si system has been studied by several investigators. Several isothermal sections have been established and are reported here.

Binary Systems

The Mo-Ni System [1991Nas] (Fig. 1) has three intermediate phases: δ (MoNi); MoNi₃; and MoNi₄. All of these phases form through peritectic or peritectoid reactions: $L + (Mo) \leftrightarrow \delta$ at 1362 °C; $(Ni) + \delta \leftrightarrow MoNi_3$ at 910 °C; and $(Ni) + MoNi_3 \leftrightarrow MoNi_4$ at 870 °C. Only one eutectic reaction, $L \leftrightarrow \delta + (Ni)$, occurs at 1317 °C.

The Mo-Si system [Massalski 2] (Fig. 2) has three intermediate phases, Mo₃Si, Mo₅Si₃, and MoSi₂, of which Mo₃Si and MoSi₂ are single-composition phases. Although the Mo₅Si₃ and MoSi₂ phases melt congruently at ~2180 and ~2050 °C, respectively, the Mo₃Si phase forms through a peritectic reaction, $L + (Mo) \leftrightarrow Mo_3Si$ at ~2025 °C. The MoSi₂ phase exists in two polymorphic forms, $\beta MoSi_2$ above 1900 °C and $\alpha MoSi_2$ below 1900 °C. Three eutectic reactions, $L \leftrightarrow Mo_3Si + Mo_5Si_3$, $L \leftrightarrow Mo_5Si_3 + \alpha MoSi_2$, and $L \leftrightarrow MoSi_2 + (Si)$, occur at 2020, ~1900, and ~1480 °C, respectively.

The Ni-Si system [1991Nas] (Fig. 3) has eight intermediate phases, Ni₃Si (β), Ni₃Si (β_2 and β_3), Ni₃₁Si₁₂ (γ'), Ni₂Si (δ'), Ni₂Si (θ), Ni₃Si₂ (ϵ and ϵ'), NiSi (ξ) and NiSi₂ (ζ and ζ'), of which several phases have polymorphic forms: β_2 and β_3 at ~25 at.% Si, ϵ and ϵ' at ~46 at.% Si, and α and β at ~66.7 at.% Si. These polymorphic transitions occur, respectively, at ~1165, a 830°, and 981 °C. The γ' , θ , and ξ phases melt congruently at 1242, 1306, and 992 °C, respectively. The β_1 , β_3 , δ' , ϵ' , and ζ' phases form through peritectic or peritectoid reactions: $L + \gamma' \leftrightarrow \beta_3$ at 1178 °C; $L + \theta \leftrightarrow \delta'$ at 1255 °C; $L + (Si) \leftrightarrow \zeta'$ at 993 °C; $(Ni) + \beta_2 \leftrightarrow \beta_1$ at 1035 °C; and $\theta + \xi \leftrightarrow \epsilon'$ at 845 °C. There are eight eutectic or eutectoid reactions in the Ni-Si system: $L \leftrightarrow \gamma + \beta_3$ at 1145 °C; $L \leftrightarrow \gamma' + \delta'$ at 1215 °C; $L \leftrightarrow \theta + \xi$ at 964 °C; $L \leftrightarrow \xi + \zeta$ at 966 °C; $\beta_2 \leftrightarrow \beta_1 + \gamma'$ at 990 °C; $\theta \leftrightarrow \delta' + \epsilon$ at 825 °C; $\epsilon' \leftrightarrow \delta' + \epsilon$ at 820 °C; $\epsilon \leftrightarrow \epsilon' + \xi$ at 800 °C; and $\epsilon \leftrightarrow \epsilon' + \delta'$ at 820 °C. The γ' , δ' , ξ , and ζ phases are single-composition phases.

Binary and Ternary Phases

There are 14 binary intermediate phases in the Mo-Ni-Si system. Several of them, however, exist in two polymorphic forms. Four ternary intermediate phases have been found to exist in the Mo-Ni-Si system. The structure data for all of the binary and ternary phases are given in Table 1.

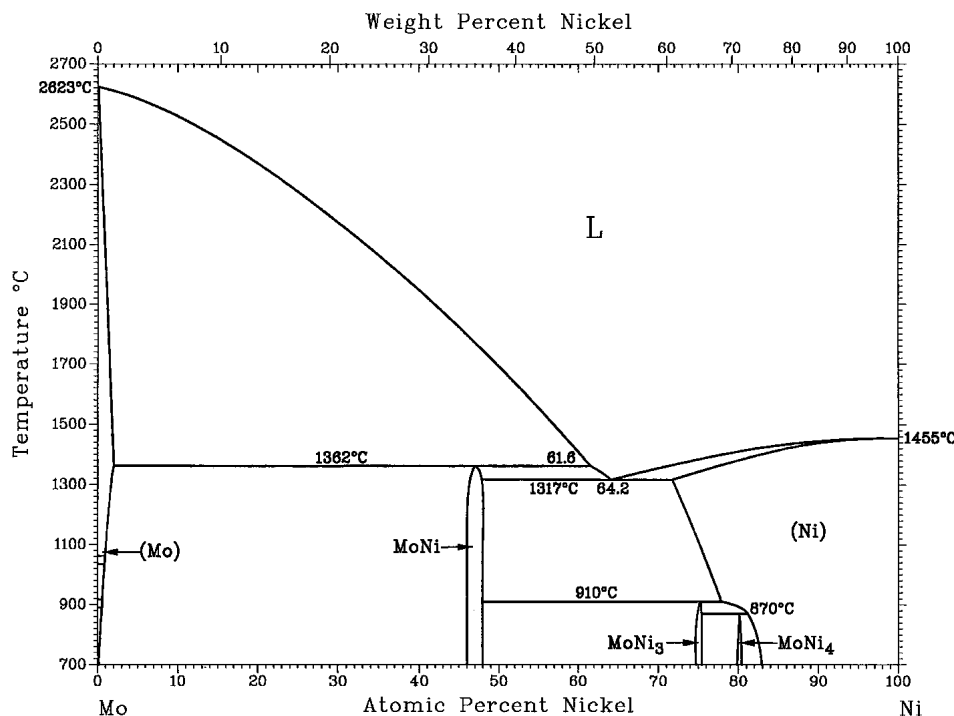


Fig. 1 Mo-Ni phase diagram [1991Nas]

Section II: Phase Diagram Evaluations

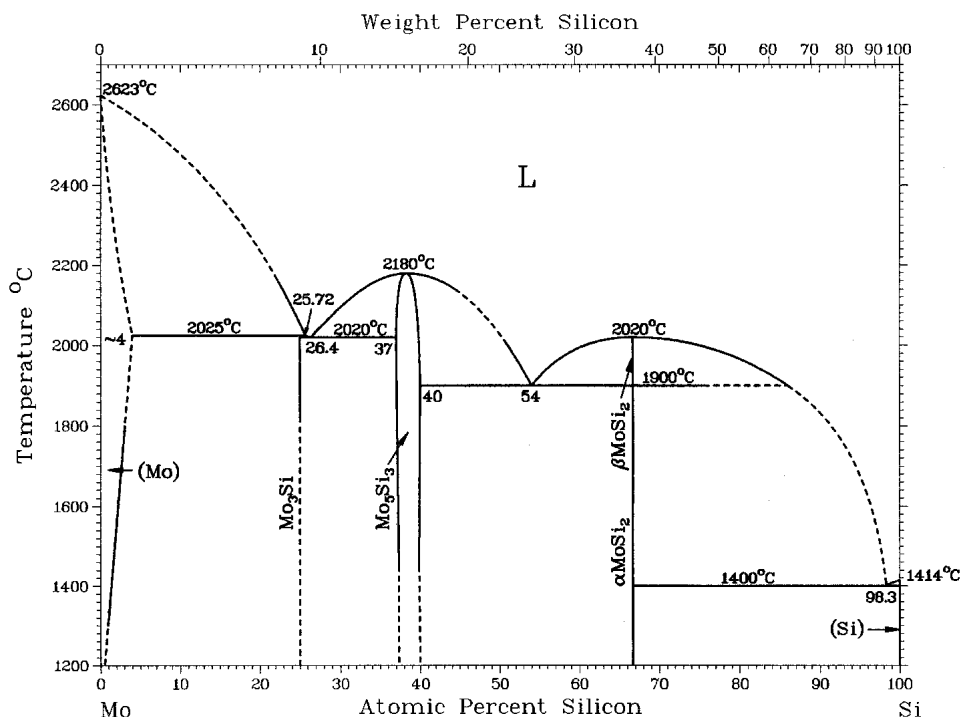


Fig. 2 Mo-Si phase diagram [Massalski 2]

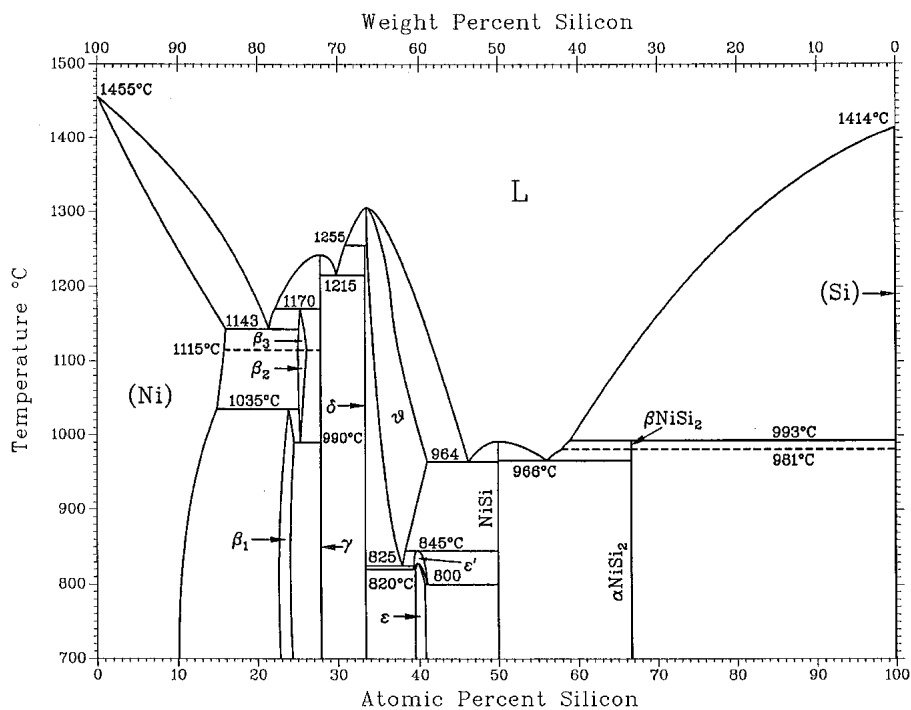


Fig. 3 Ni-Si phase diagram [1991Nas]

Ternary System

An early investigation [1925Pfa] of the high Ni region of the Mo-Ni-Si system at 700 °C has been quoted by several later investigators [1959Gua, 1969Vir]. The existence of a

ternary intermediate phase was reported by [1925Pfa]. However, [1959Gua] suggested that the Ni solid-solution boundary line as well as the position of the reported ternary intermediate phase were not determined accurately. Hence, [1959Gua] studied the Ni-rich region of the Mo-Ni-Si sys-

Table 1 Structure data of binary and ternary phases of the Mo-Ni-Si system

| Phase designation | Composition(a) | Pearson symbol | Space group | Type | Lattice parameter, nm | | |
|-------------------|--------------------------------------------------------|----------------|--------------------------------------------------------------------|-----------------------------------|-----------------------|---------------------|-------------|
| | | | | | <i>a</i> | <i>b</i> | <i>c</i> |
| γ | (Ni) | <i>cF4</i> | <i>Fm</i> $\bar{3}m$ | Cu | ... | ... | ... |
| α | (Mo) | <i>cI2</i> | <i>Im</i> $\bar{3}m$ | W | ... | ... | ... |
| Si | (Si) | <i>cF8</i> | <i>Fd</i> $\bar{3}m$ | C(diamond) | ... | ... | ... |
| ν | MoNi ₄ | <i>tI10</i> | <i>I4/m</i> | MoNi ₄ | 0.5720 | ... | 0.3564 |
| κ | MoNi ₃ | <i>oP8</i> | <i>Pmn</i> | βCu ₃ Ti | 0.5064 | 0.4224 | 0.4448 |
| δ | MoNi | <i>oP56</i> | <i>P2</i> ₁ <i>2</i> ₁ <i>2</i> ₁ | δ(MoNi) | 0.9108 | 0.9108 | 0.8852 |
| ρ | Mo ₃ Si | <i>cP8</i> | <i>Pm</i> $\bar{3}m$ | Cr ₃ Si | 0.489 | ... | ... |
| τ | Mo ₅ Si ₃ | <i>tI38</i> | <i>I4/mcn</i> | W ₅ Si ₃ | 0.728 | ... | 0.500 |
| π | αMoSi ₂ | <i>tI6</i> | <i>I4/mmm</i> | MoSi ₂ | 0.320 | ... | 0.787 |
| π' | βMoSi ₂ | ... | <i>C6</i> ₂ <i>2</i> | ... | ... | ... | ... |
| β ₁ | Ni ₃ Si(22.8-24.5) | <i>cP4</i> | <i>Pm</i> $\bar{3}m$ | AuCu ₃ | 0.350 | ... | ... |
| β ₂ | Ni ₃ Si(24.5-25.5) | <i>mC16</i> | ... | GePt ₃ | 0.697 | 0.625 | 0.507 |
| β ₃ | Ni ₃ Si(24.5-25.5) | <i>mC16</i> | ... | ... | 0.704 | β = 48.74° 0.626 | 0.508 |
| γ' | Ni ₃₁ Si ₁₂ | <i>hP43</i> | <i>P321</i> | Ni ₃₁ Si ₁₂ | 0.667 | ... | 1.228 |
| δ' | Ni ₂ Si(33.3) | <i>oP12</i> | <i>Pnma</i> | Co ₂ Si | 0.706 | 0.499 | 0.372 |
| θ | Ni ₂ Si(33.4-41.0) | <i>hP6</i> | <i>P6</i> ₃ <i>/m</i> | Ni ₂ Si | 0.3805 | ... | 0.489 |
| ε | Ni ₃ Si ₂ | <i>oP8</i> | ... | ... | ... | ... | ... |
| ε' | Ni ₃ Si ₂ | ... | ... | ... | ... | ... | ... |
| ξ | NiSi | <i>oP8</i> | <i>Pnma</i> | MnP | 0.562 | 0.518 | 0.334 |
| ζ | NiSi ₂ | <i>cF12</i> | <i>Fm</i> $\bar{3}m$ | CaF ₂ | 0.5406 | ... | ... |
| ζ' | NiSi ₂ | ... | ... | ... | ... | ... | ... |
| Γ ₁ | Mo _{33.3} Ni ₅₀ Si _{16.7} | <i>hP12</i> | <i>P6</i> ₃ <i>/mmc</i> | MgZn ₂ | 0.4709 | ... | 0.7655 |
| Γ ₁ ' | Mo ₃₂ Ni _{50.5} Si _{17.5} | ... | ... | (b) | ... | ... | ... |
| Γ ₁ '' | Mo ₃₂ Ni ₃₈ Si ₃₀ | ... | ... | (b) | ... | ... | ... |
| μ | Mo ₅₀ Ni ₄₀ Si ₁₀ | <i>R13</i> | <i>R</i> $\bar{3}m$ | W ₆ Fe ₇ | 0.4731 | ... | 2.570(c)(d) |
| R | Mo ₅₅ Ni ₄₂ Si ₃ | <i>R53</i> | <i>R</i> $\bar{3}$ | R(Mo,Co,Cr) | 1.102 | ... | 1.963(c)(d) |

(a) Composition of phases in at.%. (b) Possibly closely related to the MgZn₂-type Laves phase. (c) Lattice parameters from [1969Vir]. (d) Lattice parameters for hexagonal cell

tem in the composition region Ni-MoNi-NiSi. Alloys were prepared by arc melting high-purity component elements (Mo 99.92 mass% purity, Si 99.98 mass% purity, and carbonyl Ni containing ≤0.1 mass% impurity) and homogenized at 1100 °C for 72 h in dry hydrogen. At the Ni corner, the alloys were also annealed at 900 °C for 144 h to determine the phase boundary of the Ni solid-solution region. Phase analysis was done by x-ray diffraction (XRD) and metallography.

The partial isothermal section at 1100 °C established by [1959Gua] is given in Fig. 4. The presence of a ternary intermediate phase, φ, was reported along the line joining MoNi and NiSi at ~28 at.% Si. The φ-phase was not identified. The φ-phase was reported to be in equilibrium with the δ, γ, β₁, γ', δ', θ, and ξ-phases. At 900 °C, the Ni solid-solution phase γ was found to have smaller solubility of Mo and Si than at 1100 °C, which is to be expected on the basis of Mo-Ni and Ni-Si binary data. The dash-dot line in Fig. 4 indicates the region of high Si content at which the alloys melted at 1100 °C. The phase equilibria shown in this region correspond to a lower temperature, possibly 900 °C.

The φ phase was later identified by [1960Gla] to be a MgZn₂-type Laves phase. In search of Laves phases stabi-

lized by Si, [1961Bar] found the alloy Mo₂Ni₃Si to be of MgZn₂-type Laves phase at 1200 °C, indicating that the Laves phase is stable at high temperatures. The lattice parameters of the Mo₂Ni₃Si Laves phase were found to be *a* = 0.47 nm and *c* = 0.747 nm.

A more complete investigation of the Mo-Ni-Si system was done by [1969Vir]. High-purity Ni (purity not indicated), 99.9⁺ mass% Mo and 99.9999 mass% Si were arc-melted in argon to prepare the alloys. The alloys were wrapped in Mo foil and sealed in evacuated quartz capsules, and were annealed for 1 week at 950 °C, the capsules were then cooled in air. Only XRD was used for the characterization of the alloys. The isothermal section of the Mo-Ni-Si system at 950 °C is given in Fig. 5. Fig. 5 shows the presence of three ternary intermediate phases Γ₁, μ, and R. The presence of the R and μ phases in the Mo-Ni-Si system was reported earlier by [1965Sko] and [1966Sko], respectively. The Γ₁ phase, an MgZn₂-type Laves phase, was found to be a narrow (~2–3 at.% wide) region extending along the 33.3 at.% Mo line from ~15 to ~35 at.% Si. The μ-phase was found to be a very narrow V-shaped region, and the R-phase region was found to be an inverted triangular region ~7 at.% Mo wide and extending from about 2 to ~16 at.% Si. The

Section II: Phase Diagram Evaluations

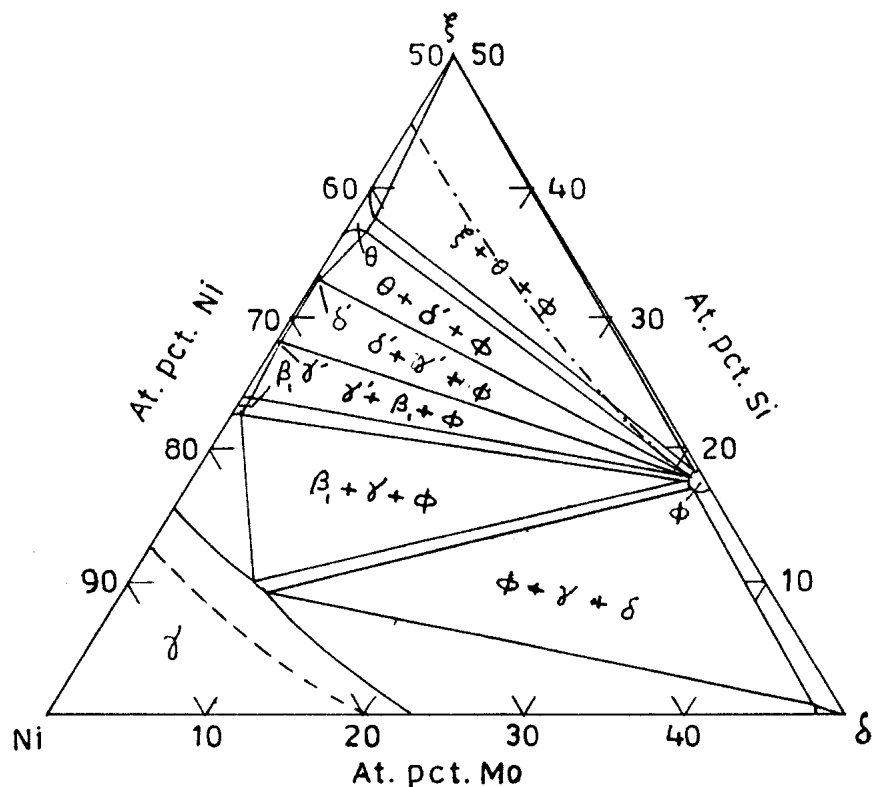


Fig. 4 A partial isothermal section of the Mo-Ni-Si system at the Ni- δ (MoNi) - ξ (NiSi) region at 1100 °C. The dash-dot line indicates the composition region of high Si content in which the alloys melted at 1100 °C. The dashed line shows the γ phase boundary at 900 °C [1959Gua].

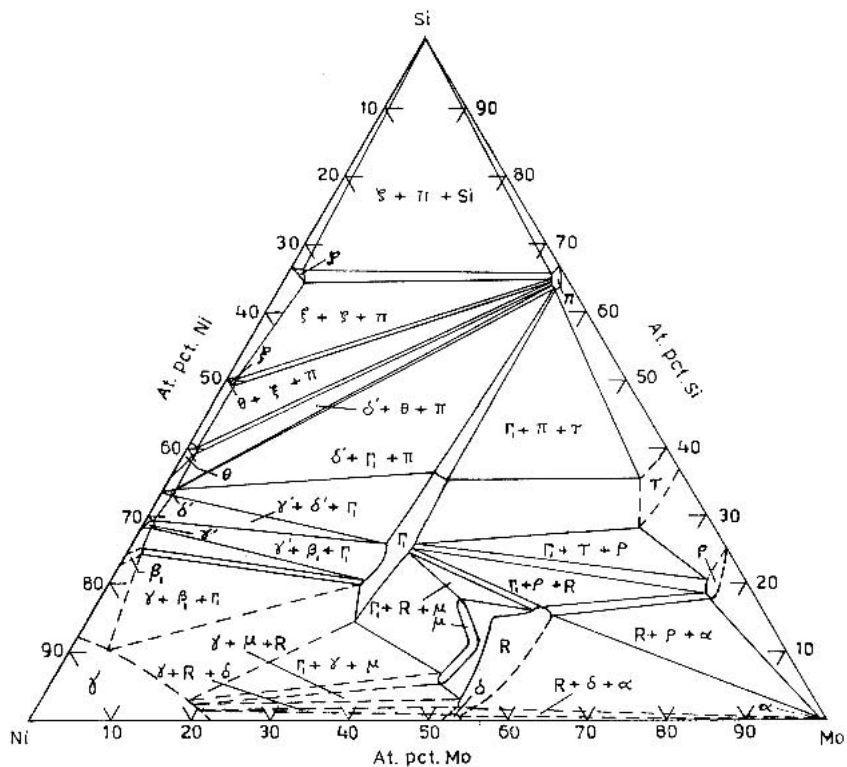


Fig. 5 The 950 °C isothermal section of the Mo-Ni-Si system [1969Vir]

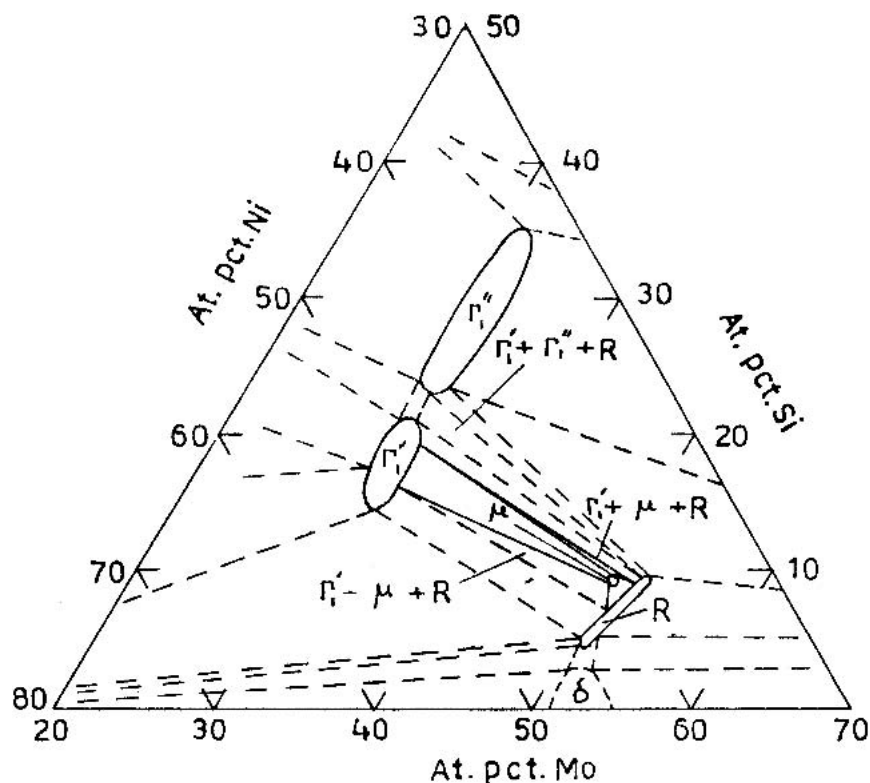


Fig. 6 A partial isothermal section of the Mo-Ni-Si system at 1280 °C [1979Sin]

MoNi δ phase did not extend to more than ~1 at.% Si. The lattice parameters of the μ and R phases were determined from two-phase alloys and are given in Table 1. The lattice parameters for the alloy with $\text{Mo}_2\text{Ni}_3\text{Si}$ composition was found to be $a = 0.4709$ nm and $c = 0.7655$ nm, which is in reasonable agreement with [1961Bar]. Unlike [1959Gua], [1969Vir] showed that at lower temperatures the Γ_1 phase is in equilibrium with γ , β_1 , γ' , δ , θ , π , τ , and ρ phases but is not in equilibrium with the ξ on Mo solid-solution phase α . In the low-Si content region, the three phase boundaries were not determined accurately, and the γ phase boundary was also not determined. The R phase was found to be in equilibrium with the γ , δ , α , μ , Γ_1 , and ρ phases, and the μ phase was found to be in equilibrium with the γ , Γ_1 , and R phases. [1969Vir] showed that all of the binary phases had some solid solubility, whereas the accepted binary data show that some of the phases are single-composition phases. Hence, in Fig. 5 the phase boundaries of some of the intermediate phases have been adjusted to conform to the binary data.

[1979Sin], in their study of crystal chemistry of complex phases present in the Mo-(Fe, Co, Ni)-Si systems, determined a partial isothermal section of the Mo-Ni-Si system at 1280 °C. The alloys were arc-melted using 99.9 mass% pure component elements near the MoNi δ phase region, up to ~40 at.% Si. The alloys were annealed at 1280 °C for 4 days in evacuated quartz capsules and quenched in cold water. Characterization of the alloys was done using metallography and x-ray diffraction. Figure 6 shows the partial isothermal section of the Mo-Ni-Si system at 1280 °C. At 1280 °C, the R-phase was found to be a very narrow (~1 at.% wide)

lens-shaped region extending from ~5 to ~9 at.% Si, and the μ phase was found to be a very small region. Two other intermediate phases, Γ_1' and Γ_1'' , were detected around the alloy compositions $\text{Mo}_{33}\text{Ni}_{50.5}\text{Si}_{17.5}$ (Γ_1') and $\text{Mo}_{32}\text{Ni}_{38}\text{Si}_{30}$ (Γ_1'') extending from ~16 to 21 at.% Si and ~23 to ~35 at.% Si, respectively. The XRD patterns of both the phases resembled closely the XRD pattern of the MgZn_2 -type Laves phase but had many extra diffraction lines that could not be accounted for on the basis of MgNi_2 -type Laves phase. Thus, the Γ_1' and Γ_1'' phases appear to be closely related to the MgZn_2 -type Laves phase. The results of [1969Vir] and [1979Sin] indicate that both the R and μ -phase regions become small as the temperature increases and suggest that these phases possibly form through peritectic reactions or they melt congruently. The MgZn_2 Laves phase was found to be stable between 900 and 1200 °C, but above 1200 °C the two MgZn_2 -related Laves phases form. Further investigation of the Mo-Ni-Si system should be done to determine how the μ and R phases form, up to what temperature the MgZn_2 and its related Laves phases remain stable, and how they form.

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